Tribological Properties and Microstructure of Electroless Nickel Coatings Reinforced with Nanoparticles

Juljeta Kaleicheva  
Faculty of Industrial Technology  
Technical University of Sofia  
Sofia, Bulgaria  
jkaleich@tu-sofia.bg

Zdravka Karaguiozova  
Space Research and Technology Institute  
Bulgarian Academy of Sciences  
Sofia, Bulgaria  
karazuzi@yahoo.com

Valentin Mishev  
Faculty of Industrial Technology  
Technical University of Sofia  
Sofia, Bulgaria  
v_mishev@tu-sofia.bg

Abstract — Composite nickel coatings composed of Ni; Ni + TiN are studied. The method for electroless nickel deposition EFFTOM-NICKEL with TiN nanosized strengthening particles (50nm) is applied. The coatings are deposited on austempered ductile iron (ADI) samples. The composition of cast iron samples is: Fe-3,63C-2,59Si-0,30Mn-0,010S-0,034P-0,53Cu wt. %. The samples are put under isothermal hardening at 900°C for an hour and isothermal retention at 290 °C for 2 hours with the aim to receive a lower bainite structure. The wear resistance experimental testing is carried out using Taber-Abraser test machine by disk to disk classical method. The microstructure observations of the coatings and padding are performed using an optical microscope GX41 OOLIMBUS also the coatings’ microhardness by Knoop Method is examined. The wear resistance, microstructure, thickness and microhardness of the as plated and thermally processed at 290°C for 6 hours coatings are defined.

Keywords — titanium nitride, electroless nickel coating, wear resistance, microhardness, lower bainite.

Introduction

The simplicity of electroless plating technology and its ability to produce high quality coatings is the reason for their popularity in surface modification and significant impact on numerous industrial applications. Nickel is a preferred metal in this method for producing coatings. Nickel coatings have excellent corrosion and wear resistance and high microhardness [1]. The necessity for the materials surface properties improvement launched the idea of the various second phase particles incorporation in the electroless nickel coatings in the 1960s [2] and led to the development of electroless nickel composite coatings. The electrolyte composite coating is formed by the co-deposition and settlement of particles on the surface of the work piece, and the subsequent envelopment of these particles by the matrix material as it is deposited. There is no molecular bonds between the particles and metal matrix [3]. The soft or hard particles are used for the co-deposition process. Several factors influence their incorporation in the electroless Ni–P matrix including, particle size and shape, relative density of the particle, particle charge, inertness of the particle, the concentration of the particles in the plating bath, the method and degree of the agitation, the compatibility of the particle with the matrix, and the orientation of the part being plated [4]. In [5] the authors outline the improvement in surface properties offered by such composite coatings and their significant impact on numerous industrial applications securing a more prominent place in the surface engineering of the metals and alloys. Thus composite coatings constitute a new class of materials which are mostly used for mechanical and tribological applications. Among these materials, nickel deposits incorporating hard ceramic particles such as silicon carbide SiC, combine anti-corrosion properties (due to the presence of nickel), with mechanical and tribological performances (due to the presence of particles of SiC). [6] studying the mechanical (hardness) and tribological (friction resistance and wear) properties of the co-deposits concludes that increasing the size or the rate of SiC particles incorporated lead to an increase in both the hardness of the films and friction coefficient when sliding against a steel ball. Author’s research [7] on the mechanism of incorporation of reinforced ZrO2, TiO2, and Al2O3 particles on 6061 aluminium alloy and on the effect of different composites on the mechanical properties of the deposit such as hardness and wear resistance proved once again that the reinforced particles as well as the heat treatment provide satisfactory improvement in hardness and wear resistance of the deposits.

Improving the properties of low-cost materials is a good opportunity to expand their application areas. Electroless nickel coatings are such an opportunity for the improvement of the cast irons surface properties [8]. The study in this work utilized electroless nickel (EN) and cathodic arc deposition (CAD) technologies with lower processing temperature to treat austempered ductile iron (ADI). The test results show that microstructures of ADI did not deteriorate after EN and CAD surface treatments. Moreover, both the EN and CAD-DLC (diamond-like carbon) coatings are identified to be amorphous type and they could be well deposited on the ADI substrate. The duplex coated DLC/EN-ADI show the highest hardness (1312 HV0.05), followed by DLC-ADI (1088 HV0.05),
EN-ADI (409 HV0.05) and then uncoated ADI (396 HV0.05). In the case of corrosion resistance, all the coated specimens are better than that of the uncoated one in 3.5wt.% NaCl aqueous solution, and the sequence is DLC/EN-ADI > EN-ADI > DLC-ADI > ADI.

In recent years, physical vapor deposition (PVD) technique using lower processing temperature has been widely adopted to coat various films, such as diamond-like carbon (DLC), CrN, TiN etc., on the engineering material for surface modification. In particular, DLC film possesses excellent mechanical properties such as high hardness and low friction coefficient. Electroless nickel (EN), another lower temperature coating process, has also a wide field of application such as the industrial components and machine parts. The purpose of the study [9] is to investigate the effect of EN and PVD–DLC surface coatings on mechanical behaviors of ADI, especially the tensile and fatigue properties.

This study is focused on the investigation of nanosized strengthening TiN particles influence on the tribological properties, microstructure and microhardness of composite nickel coatings, deposited on austempered ductile iron (ADI) samples, copper alloyed.

**Materials and Methods**

**A. Materials and heat treatment**

The EFTTOM-NICKEL technology for electroless nickel plating developed at TU of Sofia [10] is applied to obtain composite coatings with nanosized strengthening TiN particles (50nm).

The composite coatings are deposited on copper alloyed austempered ductile iron (ADI) samples. The composition of cast iron samples is: Fe-3.63C-2.59Si-0.30Mn-0.010S-0.034P-0.53Cu wt.%. The cast iron samples are put under prior to the plating process. The heat treatment consists of heating at 900°C for an hour and subsequent isothermal retention at 290°C for 2 hours. The result is to obtain ductile cast iron with a lower bainite structure (Fig. 1).

Two types of coatings are investigated: electroless nickel coating Ni and composite nickel coating with nanosized titanium nitride Ni+TiN (Table 1).

Some of the samples are heat treated at 290°C for 6 hours after coating deposition for the coatings’ adhesion improvement and microhardness increase.

The microstructure of the padding and the coatings and the coatings’ thickness are defined by means of an optical metallographic microscope GX41 OLIMPUS. The microhardness testing of the coatings is examined by Knoop method under 20g load (Table1).

**Table 1 Composition, heat treatment, microhardness and thickness of coatings**

<table>
<thead>
<tr>
<th>№</th>
<th>Composition</th>
<th>Heat treatment</th>
<th>Microhardness HK0.02</th>
<th>Thickness, [µm]</th>
</tr>
</thead>
<tbody>
<tr>
<td>1</td>
<td>Ni</td>
<td>-</td>
<td>538</td>
<td>10</td>
</tr>
<tr>
<td>2</td>
<td>Ni</td>
<td>290°C, 6h</td>
<td>930</td>
<td>10</td>
</tr>
<tr>
<td>3</td>
<td>Ni + TiN</td>
<td>-</td>
<td>588</td>
<td>8</td>
</tr>
<tr>
<td>4</td>
<td>Ni + TiN</td>
<td>290°C, 6h</td>
<td>1020</td>
<td>8</td>
</tr>
</tbody>
</table>

**B. Device and method for wear resistance testing**

The experiments on the wear resistance of nickel coatings are carrying out using a classic design “back-to-back” disc on a TABER ABRASER test machine 503, modified in accordance with the developed by the authors’ method. The device design is shown on Fig. 2.

The ring shaped sample 1 (solid) with a coating 2 is fixed on a horizontal disc 3, which is moved by an electric motor 4 with a constant angular speed w. The antibody 5 is a disc from a special abrasive material CS10. The desire normal load P in the contact surface is set through a mounted in the antibody axle, which is operated by a special device. In this way the body 1 and antibody 5 are fixed on two cross axes. Upon the constant angular speed w=const of the sample 1 and upon constant nominal contact pressure Pa=const the friction in the contact surface K keeps a constant rotation speed of the antibody 5.

![Fig.2. TABER-ABRASER test machine](image)

The test method description:

- All samples with the same ring shape and size, before the coating process are subjected to a mechanical treatment, namely grinding and polishing to ensure an equal surface roughness Ra=0.4µm. This is a binding requirement for the reliability of the wear testing, because the electroless coatings “copy” the samples surface by the plating process.
- When choosing the integral parameter “massive wear” the weight of the sample is weighed before
and after a determinate number of the disc rotation by an analytical balance WPS 180/C/2 précised to 0.1mg. The samples are treated with a special solution to neutralize the static electricity before the weighting.

- Sample 1 is fixed on a horizontal disc 3 and by the lever system the desire normal load P is set. The friction road L is determined by the number of cycles N, accounted with a cyclometer 7.

**Test basic parameters are:**

- absolute massive wear \( m \) [mg] - this is a coating lost weight in the process of the wear, estimated as a difference between the samples weight before and after the appointed number of friction cycles.
- speed of massive wear \( \dot{m} \) [mg/min] - the lost weight of the coating for a minute.
- intensity of wear \( i \) - this is the lost coating thickness for an one friction cycle. The result is a dimensionless number, which could be calculated by the formula having in mind the lost weight:

\[
i = \frac{m}{\rho Aa L}
\]  

(1)

where:

- \( \rho \) is the coating density - \( \rho=7.8 \times 10^3 \) [kg/m3];
- \( Aa \) is the nominal interaction contact surface - \( Aa =26.10^{-6} \) [m2];
- \( L \) is the friction road, estimated by the number of cycles N:

\[
L=2.\pi.R.N
\]  

(2)

where:

- \( R \) is the distance between the rotation axis of the bearing disc and the mass center of the contact place between the sample 1 and the contra body 5 (Fig. 1).

- absolute wear resistance \( I \) - this is a dimensionless number and is determined as a reciprocal value of the wear intensity, namely

\[
I = \frac{1}{i} = \frac{\rho Aa L}{m}
\]  

(3)

- nominal contact pressure \( P_a \) [N/cm²] is the normal load, per the contact interaction surface \( Aa \), i.e.

\[
P_a = \frac{P}{Aa}
\]  

(4)

- comparative index of wear resistance \( \varepsilon \) of what means the ratio between the wear resistance of the tested sample \( I_i \) and the wear resistance of a sample-standard \( I_e \), i.e it is a dimensionless number, indicating how many times the wear resistance of the tested sample is higher compared to the sample-standard under the same contact interaction conditions:

\[
\varepsilon_{ie} = \frac{I_i}{I_e}
\]  

(5)

The parameters of the contact interaction are presented in Table 2.

<table>
<thead>
<tr>
<th>Table 2 Test Parameters</th>
</tr>
</thead>
<tbody>
<tr>
<td>Nominal contact surface</td>
</tr>
<tr>
<td>( A_a ) [m²]</td>
</tr>
<tr>
<td>26.10⁻⁴</td>
</tr>
<tr>
<td>Nominal contact pressure</td>
</tr>
<tr>
<td>( P_a ) [N/cm²]</td>
</tr>
<tr>
<td>47.15</td>
</tr>
<tr>
<td>Average speed of sliding</td>
</tr>
<tr>
<td>( V ) [cm/s]</td>
</tr>
<tr>
<td>17.90</td>
</tr>
</tbody>
</table>

**RESULTS AND DISCUSSION**

The coatings’ thickness is between 8÷10μm (Table 1). The microhardness HK0,02 test results of the coatings are presented in Table 1. The heat treatment of the coatings at 290°C for 6 hours leads to the coatings’ microhardness twice increase.

Electroless Ni-coatings show amorphous structure in an as-plated state. The coatings’ structure becomes crystalline after heat treatment at 290°C, 6h. The diffraction patterns of the samples prove the presence of Ni3P phase in the coatings’ structure [12]. The heat treatment at 290°C, 6h leads to a crystal formation of Ni3P phase, which is dispersed and increases the microhardness of the coatings [11], [12]. The introduced nano- and microparticles in coatings deposited by various methods further increase their microhardness [13].

The coatings’ microstructure appears as a white strip following the sample relief (Figs 3 and 4). The padding microstructure of the cast iron, obtained at a low temperature isothermal retention in the bainitic field consists of a lower bainite (Figs 1, 3 and 4).
The experimental results received for the massive wear rate in dependence on the friction road L for the different coatings are presented in Table 4.

**Table 4 Massive wear rate variation in friction road**

<table>
<thead>
<tr>
<th>№</th>
<th>m [mg/min]</th>
<th>Wear intensity i</th>
<th>Wear resistance I</th>
</tr>
</thead>
<tbody>
<tr>
<td>1</td>
<td>8.9</td>
<td>2.72.10^-7</td>
<td>0.37.10^-7</td>
</tr>
<tr>
<td>2</td>
<td>10.5</td>
<td>3.2.10^-7</td>
<td>0.31.10^-7</td>
</tr>
<tr>
<td>3</td>
<td>6.9</td>
<td>2.11.10^-7</td>
<td>0.47.10^-7</td>
</tr>
<tr>
<td>4</td>
<td>5.2</td>
<td>1.6.10^-7</td>
<td>0.63.10^-7</td>
</tr>
</tbody>
</table>

The analysis of the wear resistance value (Fig. 5) shows the sample 4 with Ni+TiN coating with heat treatment at 290°C, 6h possesses higher wear resistance (I = 0.63.10^-7) than this one of the sample 3 with the same coating but without heat treatment (I = 0.47.10^-7). The measured higher microhardness of the Ni+TiN coating with heat treatment at 290°C, 6h (1020 HK0,02) compared to this one of the same coating without heat treatment (588 HK0,02) corresponds to higher wear resistance (Table 1 and Fig. 5). The lack of correlation between the microhardness and wear resistance of the coatings of pure Ni (samples 1 and 2) could be attributed to the low adhesion between the coating and the padding. This could be explained with the graphite presence in the iron microstructure.

**Table 5 Massive wear variation in time**

<table>
<thead>
<tr>
<th>№</th>
<th>2,5 [min]</th>
<th>5 [min]</th>
<th>7,5 [min]</th>
<th>10 [min]</th>
<th>12,5 [min]</th>
<th>15 [min]</th>
</tr>
</thead>
<tbody>
<tr>
<td>1</td>
<td>3,2</td>
<td>6,0</td>
<td>6,7</td>
<td>7,9</td>
<td>8,9</td>
<td></td>
</tr>
<tr>
<td>2</td>
<td>3,8</td>
<td>4,7</td>
<td>6,9</td>
<td>8,0</td>
<td>9,8</td>
<td>10,5</td>
</tr>
<tr>
<td>3</td>
<td>2,7</td>
<td>3,8</td>
<td>4,3</td>
<td>5,3</td>
<td>6,0</td>
<td>6,9</td>
</tr>
<tr>
<td>4</td>
<td>1,5</td>
<td>2,8</td>
<td>4,0</td>
<td>4,6</td>
<td>4,9</td>
<td>5,2</td>
</tr>
</tbody>
</table>

The experimental results received for the massive wear rate m dependence on the friction road I for the different coatings are presented in Table 3.

**Table 3 Massive wear variation in time**

**REFERENCES**